



Hydrodeamination of β -enamino ketones to 1,2-dideoxy-D-threo-3-hexulose via palladium

Zi-Ping Lin^a, Hui-Chang Lin^{a,*}, Hsu-Hsuan Wu^a, Hsiu-Wen Chou^a, Shao-Kai Lin^b, Kuan-Chin Sung^c, Fung Fuh Wong^a

^a Graduate Institute of Pharmaceutical Chemistry, China Medical University, No. 91, Hsueh-Shih Rd., Taichung, Taiwan 40402, Taiwan, ROC

^b Sustainable Environment Research Center, National Cheng Kung University, No. 500, Sec. 3, An-ming Rd., Tainan City, Taiwan 709, Taiwan, ROC

^c Department of Surgery, Attending Physician Neurosurgery, No. 201, Taikang Village, Liouying Township, Taiwan 709, Taiwan, ROC

ARTICLE INFO

Article history:

Received 27 April 2009

Revised 19 June 2009

Accepted 23 June 2009

Available online 30 June 2009

Keywords:

Hydrodeamination
 β -Enamino ketones
 β -Mannosidase
 Hex-1-en-3-ulose

ABSTRACT

β -Enamino ketones were successfully synthesized in good to excellent yields by reaction of hex-1-en-3-uloses with amines. After hydrogenation on palladium catalyst, β -enamino ketones effectively underwent hydrodeamination and were converted to the corresponding 1,2-dideoxy-D-threo-3-hexulose derivatives in 89–95% yields.

Crown Copyright © 2009 Published by Elsevier Ltd. All rights reserved.

1. Introduction

In 2007, Wong and co-workers reported the (2*R*,3*R*,4*R*)-2-(hydroxymethyl)pyrrolidine-3,4-diol compound **1** as quite specific and slightly more potent inhibitor of β -mannosidase.¹ (4*S*,5*R*)-6-Azido-4,5-dihydroxy-hexan-3-one (**2**)¹ and 1,2-dideoxy-D-threo-3-hexulose (**3**)² can be used as the precursors to prepare **1** (see Fig. 1). Herein, we report an efficient and convenient methodology to synthesize **3** from hex-1-en-3-ulose starting materials.

Hydrodeamination³ was widely applied in the field of organic functional group transformation, including diazotization,⁴ amination of the toluenesulfonamide,⁵ nitrosation,⁶ difluoroamine,^{3b} and nucleophilic aromatic substitutions.⁷ Many hydrodeaminations of benzylic amines, benzoyl amines, or primary amines have been investigated by catalytic hydrogenation, dissolving metal reduction,⁸ or other reducing agents.⁹ There have been a few directed hydrodeamination methods reported for the conversion of β -enamino ketones to ketones.¹⁰ In this work, we reported the novel hydrodeamination of β -enamino ketones by hydrogenation on palladium catalyst to generate the corresponding 1,2-dideoxy-D-threo-3-hexulose products.

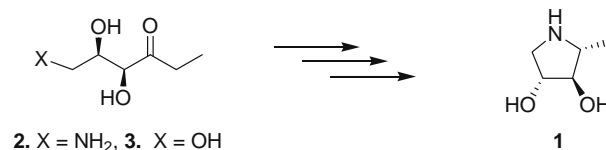


Figure 1.

2. Results and discussion

β -Enamino ketones **9–21** were prepared as the reference materials following the previous publication procedure via Michael addition of hex-1-en-3-uloses.¹¹ Hex-1-en-3-uloses **4–8** reacted with two equivalents of primary amines, including benzylamine, *n*-butylamine, and *n*-octylamine in MeOH for two hours to give the corresponding β -enamino ketones **9–21** in 80–96% yields (see Table 1).¹¹ The addition reactions were clean and smooth to give the high diastereoselective *Z*-geometry of the products due to the intramolecular hydrogen bonding.¹² All of β -enamino ketone structures were identified and determined by DEPT, NOESY, and other spectroscopic methods. For example, compound **10** possessed a characteristic broad singlet resonance at δ 10.11 for the amino proton, a doublet doublet resonance at δ 6.81 ($J = 7.2, 13$ Hz) for the alkene proton on C₁, and a doublet at δ 5.27 ($J = 7.2$ Hz) for the alkene proton on C₂.

* Corresponding author. Tel.: +886 4 2205 3366x5612; fax: +886 4 2207 8083.
 E-mail addresses: lhc550005@yahoo.com.tw, huichang@mail.cmu.edu.tw (H.-C. Lin).

Table 1

The results of Michael addition of hex-1-en-3-uloses

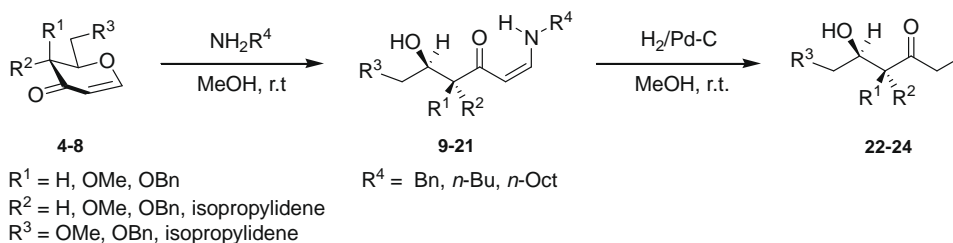
Hex-1-en-3-uloses (4–8)				β-Enamino ketones 9–21		
Substrates	R ¹	R ²	R ³	Compounds	R ⁴	Yields (%)
4	H	OMe	OMe	9	Bn	80
4	H	OMe	OMe	10	<i>n</i> -Bu	97
4	H	OMe	OMe	11	<i>n</i> -Oct	84
5	OMe	H	OMe	12	Bn	91
5	OMe	H	OMe	13	<i>n</i> -Bu	96
5	OMe	H	OMe	14	<i>n</i> -Oct	95
6	H	Isopropylidene		15	Bn	92
6	H	Isopropylidene		16	<i>n</i> -Bu	89
6	H	Isopropylidene		17	<i>n</i> -Oct	80
7	H	OBn	OBn	18	Bn	93
8	H	OBn	OBn	19	<i>n</i> -Bu	90
8	OBn	H	OBn	20	Bn	95
8	OBn	H	OBn	21	<i>n</i> -Bu	89

Palladium catalyst was commonly applied in the hydrogenation of organic compounds.¹³ In the newly developed hydrodeamination, various β-enamino ketones **9–17** were treated with a catalytic amount of palladium in MeOH at room temperature for 4 h through the atmosphere by a hydrogen-filled balloon. After the reactions were complete, the normal work-up and purification with column chromatography on silica gel were performed. The desired 1,2-dideoxy-D-3-hexuloses were isolated in liquid form (**22–24**, see Scheme 1 and Table 2).

The benzylic group is one of the most popular protecting units for alcohol and carbohydrate compounds.¹⁴ By applying the standard hydrodeamination procedure to benzyl-protected β-enamino ketones **18** and **19** (R¹ = H, R² = R³ = OBn), the O-deprotected and hydrodeaminated 1,2-dideoxy-D-threo-3-hexulose **25** (R¹ = H, R² = R³ = OH) was afforded in 89% and 93% yield, respectively. Furthermore, we performed the hydrodeamination to benzyl-protected β-enamino ketones **20** and **21**, in which the amino moiety was linked with Bn and *n*-Bu groups on nitrogen atom. The corresponding 1,2-dideoxy-D-erythro-3-hexulose **26** was produced in 89% and 95% yield, respectively, (see Scheme 2). Comparing benzyl-protected β-enamino ketones (**18–21**) with methyl-protected β-enamino ketones (**9–17**), we found that the benzyl-protected group was easily hydrogenated to hydroxyl group under the same reaction condition.

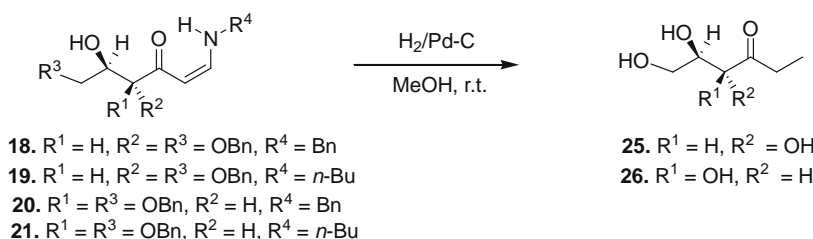
When we treated hex-1-en-3-uloses **5** with *N,N*-diethylamine (secondary amine) and acetamide by following the previous procedure, only the corresponding *N,N*-diethyl-β-enamino ketone **27** was obtained in 87% yield (see Scheme 3). We applied the same hydrodeamination to *N,N*-diethyl-β-enamino ketone **27**. The hydrogenated product **23** also succeeded in generating 90% yield.

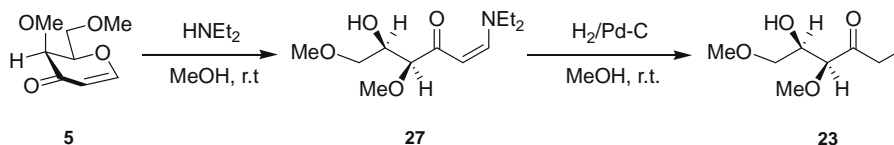
We proposed a plausible mechanism for the hydrodeamination of β-enamino ketones **28** to 1,2-dideoxy-D-threo-3-hexulose **33** as shown in Scheme 4, which accounted for our approach and design. Palladium catalyst underwent hydrogenation toward β-enamino ketone **28** to give hydrido olefin palladium complex **29** and isomer

**Scheme 1.****Table 2**

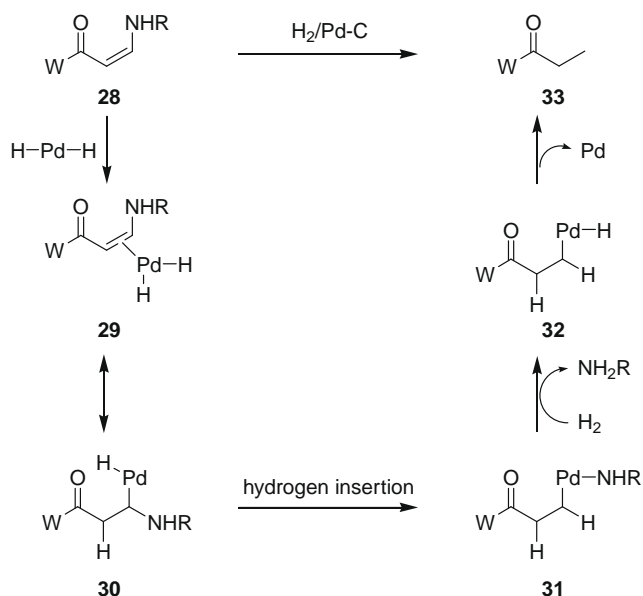
The results of hydrodeamination of β-enamino ketones 9–17

Substrates	β-Enamino ketones 9–21				1,2-Dideoxy-D-threo-3-hexuloses 22–24	
	R ¹	R ²	R ³	R ⁴	Compounds	Yields (%)
9	H	OMe	OMe	Bn	22	95
10	H	OMe	OMe	<i>n</i> -Bu	22	92
11	H	OMe	OMe	<i>n</i> -Oct	22	94
12	OMe	H	OMe	Bn	23	90
13	OMe	H	OMe	<i>n</i> -Bu	23	96
14	OMe	H	OMe	<i>n</i> -Oct	23	89
15	H		Isopropylidene	Bn	24	93
16	H		Isopropylidene	<i>n</i> -Bu	24	92
17	H		Isopropylidene	<i>n</i> -Oct	24	90

**Scheme 2.**



Scheme 3.



Scheme 4.

30.¹⁵ The intramolecular hydrogen insertion took place to give aminopalladation adduct **31**.¹⁵ After further hydrogenation, the intermediate **32** and by-product amine (NH_2R) were generated. In a control experiment for the hydrodeamination, we were able to identify the resultant amine (NH_2R) as a by-product by GC–mass spectroscopic technique. Finally, hydrogenation of intermediate **32** gave the target 1,2-dideoxy-D-threo-3-hexulose **33**.

In conclusion, β -enamino ketones were prepared as the starting materials by Michael addition of hex-1-en-3-uloses with primary amines. They underwent the novel hydrodeamination with a catalytic amount of palladium under hydrogen atmosphere to give 1,2-dideoxy-D-threo-3-hexuloses in excellent yields (89–93%).

Acknowledgments

The authors thank the National Science Council of Taiwan (NSC95-2314-B-039-046 and NSC96-2113-M-039-002) and China Medical University, Taiwan for their financial support. The authors also wish to thank Dr. C. H. Lin and Prof. C. Y. Wu for fruitful discussion of the work.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at doi:10.1016/j.tetlet.2009.06.112.

References and notes

- Sugiyama, M.; Hong, Z.; Liang, P. H.; Dean, S. M.; Whalen, L. J.; Greenberg, W. A.; Wong, C. H. *J. Am. Chem. Soc.* **2007**, *129*, 14811.
- Lehmann, J.; Petry, S. *Carbohydr. Res.* **1993**, *239*, 133.
- (a) Doldouras, G. A.; Koionitsch, J. *J. Am. Chem. Soc.* **1978**, *100*, 341; (b) Bumgardner, C. L.; Martin, K. J.; Freeman, J. P. *J. Am. Chem. Soc.* **1963**, *85*, 97; (c) Nickon, A.; Hill, A. S. *J. Am. Chem. Soc.* **1964**, *86*, 1152; (d) Hutchins, R. O.; Cistone, F.; Goldsmith, B.; Heumnu, P. *J. Chem.* **1976**, *40*, 2018; (e) Kntritzky, A. R.; Horvath, K.; Plau, B. *J. Chem. Soc., Perkin Trans. 1* **1980**, 2254; (f) Barton, D. H. R.; Bringmann, G.; Lamotte, G. *Tetrahedron Lett.* **1979**, *24*, 2291; (g) Baumgarten, R. J.; Curtis, V. A. In *The Chemistry of Amino, Nitro, and Nitro Compounds and Their Derivatives, Part 2*; Patai, S., Ed.; John Wiley and Sons: New York, 1982; p 929; A related reductive transformation of 1,2-disubstituted tosylhydrazinea has also been reported: (h) Cram, D. S.; Bradshaw, J. S. *J. Am. Chem. Soc.* **1963**, *85*, 3525.
- (a) Tsurumi, K.; Abe, A.; Fujimura, H.; Asai, H.; Nagasaka, M.; Mikaye, H. *Folia Pharmacol. Jpn.* **1976**, *72*, 41; (b) Rapposelli, S.; Lapucci, A.; Minutolo, F.; Orlandini, E.; Ortore, G.; Pinza, M.; Balsamo, A. *Farmaco* **2004**, *59*, 25; (c) Souza, F. R.; Souza, V. T.; Ratzlaff, V.; Borges, L. P.; Oliverira, M. R.; Bonacorso, H. G.; Zanatta, N.; Martins, M. A. P.; Mello, C. F. *Eur. J. Pharmacol.* **2002**, *451*, 141; (d) Ek, F.; Axelsson, O.; Wistrand, L. G.; Frejd, T. *J. Org. Chem.* **2002**, *67*, 6376; (e) Ek, F.; Axelsson, O.; Wistrand, L. G.; Frejd, T. *J. Org. Chem.* **2003**, *68*, 1911; (f) Butler, R. N. *Chem. Rev.* **1975**, *75*, 241; (g) Maslak, P.; Fanwick, P. E.; Guthrie, R. D. *J. Org. Chem.* **1984**, *49*, 655.
- (a) Guziec, F. S., Jr.; Wei, D. *J. Org. Chem.* **1992**, *57*, 3772; (b) Wang, Y.; Guziec, F. S., Jr. *J. Org. Chem.* **2001**, *66*, 8293; (c) Guziec, F. S., Jr.; Wei, D. *Tetrahedron Lett.* **1992**, *33*, 7465.
- (a) White, E. H. *J. Am. Chem. Soc.* **1954**, *76*, 4497; (b) Doyle, M. P.; Dellaria, J. F., Jr.; Siegfried, B.; Bishop, S. W. *J. Org. Chem.* **1977**, *42*, 3494.
- Basaif, S. A.; Hassan, M. A.; Gobouri, A. A. *Dyes Pigments* **2007**, *72*, 387.
- (a) du Vigneaud, V.; Behrens, O. K. *J. Biol. Chem.* **1937**, *117*, 27; (b) Wan, X.; Xing, D.; Fang, Z.; Li, B.; Zhao, F.; Zhang, K.; Yang, L.; Shi, Z. *J. Am. Chem. Soc.* **2006**, *128*, 12046.
- (a) Coulter, J. M.; Lewis, J. W.; Lynch, P. P. *Tetrahedron* **1968**, *24*, 4489; (b) Singaram, B.; Goralski, C. T.; Rangaishenvi, M. V.; Brown, H. C. *J. Am. Chem. Soc.* **1989**, *111*, 384.
- Kotschetkow, Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya; 1954, 47, 52.
- Paulsen, H.; Buensch, H. *Chem. Ber.* **1978**, *111*, 3484.
- (a) Pennington, F. C.; Kehret, W. D. *J. Org. Chem.* **1967**, *32*, 2034; (b) Kashima, C.; Aoyama, H.; Yamamoto, Y.; Nishi, T.; Yamada, K. *J. Chem. Soc., Perkin Trans. 2* **1975**, 665.
- (a) Studer, M.; Blaser, H.-U.; Exner, C. *Adv. Synth. Catal.* **2003**, *345*, 45; (b) Baiker, A. *J. Mol. Catal. A: Chem.* **2000**, *163*, 205; Communications: (c) Glorius, F.; Spielkamp, N.; Holle, S.; Goddard, R.; Lehmann, C. W. *Angew. Chem., Int. Ed.* **2004**, *43*, 2850; (d) Mhadgut, S. C.; Bucsi, I.; Toeroek, M.; Toeroek, B. *Chem. Commun.* **2004**, 984; (e) Raynor, S. A.; Thomas, J. M.; Raja, R.; Johnson, B. F. G.; Bell, R. G.; Mantle, M. D. *Chem. Commun.* **2000**, 1925.
- (a) Freedman, H. H.; Dubois, R. A. *Tetrahedron Lett.* **1975**, 3251; (b) Bouzide, A.; Sauvé, G. *Tetrahedron Lett.* **1977**, *38*, 5945; (c) Yamashita, M.; Takegami, Y. *Synthesis* **1977**, 803; (d) Fukuzawa, A.; Sato, H.; Masamune, T. *Tetrahedron Lett.* **1987**, *28*, 4303; (e) Cruzado, C.; Bernabe, M.; Martin-Lomas, M. *J. Org. Chem.* **1989**, *54*, 465; (f) Classon, B.; Garegg, P. J.; Oscarson, S.; Tidén, A. K. *Carbohydr. Res.* **1991**, *216*, 187.
- Keinan, E.; Greenspoon, N. *J. Am. Chem. Soc.* **1986**, *108*, 7314.